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Combining photochemistry and catalysis: rapid access to sp^3 – rich polyheterocycles from simple pyrroles†

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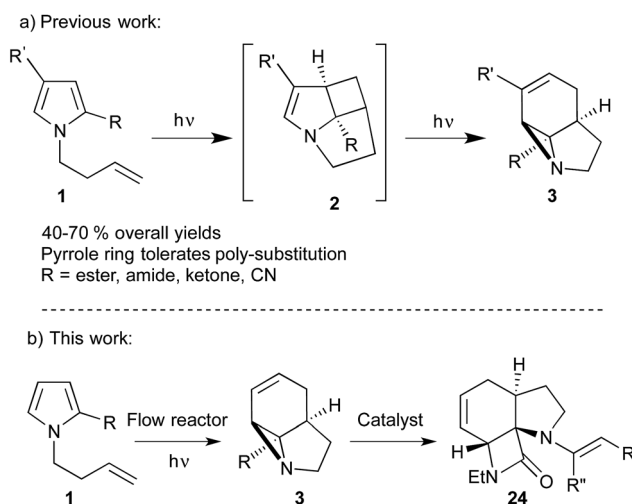
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Use of FEP flow reactor technology allows access to gram quantities of photochemically-generated tricyclic aziridines. These undergo a range of novel palladium-catalyzed ring-opening and cycloaddition reactions, likely driven by their inherent strain, allowing incorporation of further functionality by fusing additional heterocyclic rings onto these already complex polycyclic cores. This rapid, 2-step access to complex sp^3 – rich heterocycles should be of interest to those in the fields of drug discovery and natural product synthesis.

Introduction

The strained 3-membered aziridine ring has long been exploited in a variety of ring-opening reactions, serving as a valuable strategy in the synthesis of amino compounds. Compared to epoxides, however, aziridines are generally less reactive due to the lower electronegativity of N vs. O. Consequently the reactivity of aziridines is usually enhanced by carrying out reactions with nucleophiles in the presence of Brønsted or Lewis acids, or by the attachment of an electron withdrawing group to the nitrogen.¹ The latter approach, however, restricts the type of aziridines that can be used and requires removal of the activating group in a subsequent step. We recently described a photochemical transformation of simple *N*-butenyl substituted pyrroles **1** into tricyclic fused aziridines **3**, via cyclobutane **2** (Scheme 1a).² However, we found the batch scale-up of this very powerful sequence to be somewhat limited. This is due to the low overall quantum yield of this 2-step, 2-photon sequence and the fact that, due to the high extinction coefficient ($\epsilon = 13\,000$), the reaction has to be run at high dilution (<0.02 mM). This made further study of the chemistry of the resulting aziridines challenging.

Herein we describe the production of multigram quantities of aziridines using fluorinated ethylene propylene (FEP) flow reactor technology, a concept that has seen increasing use in recent years.³ The resulting structural complexity and close



Scheme 1 (a) Formation of aziridines by photocycloaddition/rearrangement of pyrrole derivatives. (b) Scale-up of photochemical aziridine synthesis and subsequent catalytic ring opening to generate polyheterocycles.

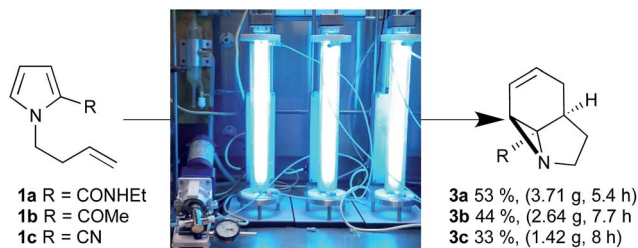
proximity of functional groups within these aziridines prompted us to explore their reactivity with various nucleophiles and ring-opening conditions. Their interesting, and sometimes unusual results are reported and include a novel β -lactam forming sequence (Scheme 1b).

Results and discussion

Initially, we examined the scale-up irradiations of the three pyrroles **1a–c** at 254 nm. This used the previously described trio of 1-layer FEP reactors with 36 W low-pressure Hg lamps.^{3b} The three reactors were connected together in series for maximum

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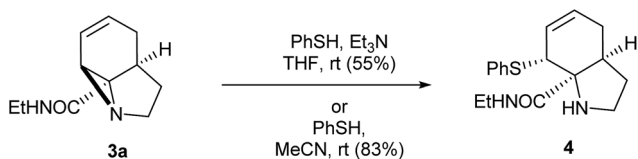


Scheme 2 Scale up of photochemical aziridine synthesis using a trio of 1-layer FEP reactors with 36 W low-pressure Hg lamps.

productivity. Once optimized this gave gram quantities of the three aziridines **3a–c** per 5–8 h run (Scheme 2).

With these quantities in hand we first explored the ring opening of the amide-aziridine **3a** with nucleophiles (Scheme 3). Using thiophenol the aziridine underwent S_N2 ring-opening under mild conditions in THF in the presence of triethylamine. The high reactivity of the unactivated aziridine ring was highlighted by the clean ring opening with thiophenol in acetonitrile without added base to give **4**. We considered that this ring-opening of an unactivated aziridine under such mild conditions pointed towards further strain imposed by the tricyclic system. Unfortunately ring opening could not be achieved using nucleophiles such as phenol or methanol under neutral or basic conditions. When the reaction was repeated with methanol under acid catalysis (*p*-TSA) an inseparable mixture of products was obtained which appeared to be a mixture of S_N2' ring-opened diastereomers. A screen of other catalysts was not carried out as the aziridine was prone to decomposition under acidic conditions.

Literature reports of Pd-catalysed Tsuji–Trost ring-opening reactions of vinyl aziridines are rare.⁴ Indeed there would appear to be only two reports involving the addition of carbon-nucleophile in an S_N2' mode.⁵ We were intrigued therefore to see if our highly strained systems would extend the scope of these reactions to a wider range of nucleophiles. Treatment of **3a** with phenol and 5 mol% $Pd(PPh_3)_4$ in dioxane gave exclusively the *anti* product **5** (entry 1, Table 1). Despite the lack of literature precedent we then examined the reaction with a range of carbon-nucleophiles. Pleasingly the reaction of **3a** and **3b** also proceeded smoothly with a selection of soft carbon-nucleophiles (entries 3–10). It was also found that the *anti/syn* ratios could be switched on moving to more polar solvents.⁶ For example, reaction of **3a** in DMF at 80 °C resulted in a 1 : 4 ratio of **5** : **6** (Nu = PhO, entry 2). Similarly reaction of **3a** in MeCN with pentane-2,3-dione provided a switch in the *anti/syn* ratio to 1 : 9 (entry 6). This switch is likely to reflect a change from an



Scheme 3 Initial ring opening reactions of aziridine **3a**.

Table 1 Ring opening of aziridines with nucleophiles under Pd(0) catalysis

Entry	R	Solvent	NuH	T (°C)	Yield (%)	5 : 6
1	CONHEt	Dioxane	PhOH	rt	49	1 : 0
2	CONHEt	DMF	PhOH	80	52	1 : 4
3	CONHEt	Dioxane	NC-CH ₂ -CN	40	52	1 : 0
4	CONHEt	MeCN	NC-CH ₂ -CN	40	57	1 : 1
5	CONHEt	Dioxane	CH ₃ -C(=O)-CH ₂ -C(=O)-CH ₃	80	82	1 : 0
6	CONHEt	MeCN	CH ₃ -C(=O)-CH ₂ -C(=O)-CH ₃	80	67	1 : 9
7	CONHEt	Dioxane	2,2,6,6-tetramethylcyclohexanone	rt	75	7 : 1
8	CONHEt	Dioxane	2,2,3,3-tetramethylindane-1,4-dione	rt	64	1 : 0
9	COMe	Dioxane	2,2,6,6-tetramethylcyclohexanone	rt	52	13 : 1
10	COMe	Dioxane	2,2,3,3-tetramethylindane-1,4-dione	rt	43	5 : 1

outer sphere mechanism^{6a,b} (direct attack of nucleophile on carbon – overall retention) to an inner sphere process^{6c} (initial attack on metal – overall inversion) on moving to a more polar solvent. Such behaviour has been observed in Tsuji–Trost reactions of allylic acetates on switching from soft to hard nucleophiles,^{6d} and appears to reflect the decrease in stability of the anion on moving to more polar solvents as judged from the pK_a values of phenol in this case.⁷

It has been previously demonstrated⁸ that *N*-alkyl substituted 2-vinyl aziridines reacted with isocyanates under Pd(0) catalysis to generate vinyl substituted cyclic ureas. If this could be applied to **3** then highly functionalised tricyclic ureas such as **7** would be accessible in just two steps from simple pyrroles. Table 2 describes the reaction between a wide range of aryl and sulfonyl isocyanates with tricyclic aziridines under Pd(0) catalysis. In general, the ureas **7** were formed in good yield using a 5 mol% loading of $Pd(PPh_3)_4$. It was found that the



Table 2 Pd-catalysed [3 + 2] cycloadditions with isocyanates

Entry	R	R ₂	Yield (%)	7 : 8
1	CONHEt	4-MeC ₆ H ₄ SO ₂	83	1 : 0
2	CN	4-MeC ₆ H ₄ SO ₂	81	1 : 0
3	COMe	4-MeC ₆ H ₄ SO ₂	43	1 : 0
4	CONHEt	4-ClC ₆ H ₄	58	1 : 0
5	CONHEt	2-ClC ₆ H ₄	71	1 : 0
6	CN	2-ClC ₆ H ₄	49	1 : 0
7	CONHEt	2-CF ₃ C ₆ H ₄	81	1 : 0
8	CONHEt	3-CF ₃ C ₆ H ₄	91	1 : 0
9	CONHEt	4-CF ₃ C ₆ H ₄	54	1 : 0
10	COMe	4-CF ₃ C ₆ H ₄	55	1 : 0
11	CONHEt	2-NO ₂ C ₆ H ₄	89	1 : 0
12	CN	2-NO ₂ C ₆ H ₄	54	1 : 0
13	CONHEt	4-NO ₂ C ₆ H ₄	89	1 : 0
14	CONHEt	4-MeCOC ₆ H ₄	79	1 : 0
15	CONHEt	4-MeOC ₆ H ₄	80	1 : 0
16	CONHEt	2-MeOC ₆ H ₄	38	1 : 0
17	CONHEt	2,6-Cl ₂ C ₆ H ₃	80	10 : 1
18 ^a	CONHEt	4-MeC ₆ H ₄ SO ₂	51	0 : 1
19 ^b	CONHEt	4-MeC ₆ H ₄ SO ₂	85	0 : 1
20 ^b	CONHEt	2-ClC ₆ H ₄	96	0 : 1

^a Uncatalysed, dioxane, rt. ^b 1 mol% Pd(Ph₃P)₄, dioxane, rt.

amide-aziridine **3a** reacted more consistently than the COMe (**3b**) or CN (**3c**) substituted derivatives. A range of aryl isocyanates were tolerated, although 2-substituted examples tended to give lower yields and 2,6-substitution lead to mixed products from N/O cyclisation (entry 17). An interesting observation was noted with low (1 mol%) catalyst loadings (entries 19 & 20) and indeed no catalyst (entry 18), where exclusive formation of the cyclic imidates **8** was observed. In the case entry 19 increasing the quantity of Pd to 5 mol% results in rapid conversion of **8** to **7**. These results are in good agreement with those observed by Alper for simpler vinyl aziridines.^{8a}

We then examined the potential [3 + 2] cycloaddition reactions of alkenes with the aziridines, as rapid access to the resulting tricyclic fused pyrrolidines would be attractive for alkaloid synthesis. Such Pd-catalysed [3 + 2] cycloadditions have been examined previously, but generally require activated aziridines.⁹ Initial results between **3a** and simple alkenes such as methyl acrylate, acrylonitrile and methyl vinyl ketone showed no reactions. After much optimisation, benzylidene malononitrile was found to react under the specific conditions shown in Table 3 to give the pyrrolidine **10a** as essentially one diastereomer. Higher yields were obtained by the use of the more reactive methylene diesters, which led to the formation of the tricyclic pyrrolidines **11a** and **12a, b**. Key to the success of these reactions was the use of triphenylphosphite as ligand;¹⁰ phosphine ligands gave lower yields and polymerisation of the

Table 3 Pd-catalysed [3 + 2] cycloadditions of aziridines with alkenes, aldehydes and imines

3a = CONHEt
3b = COMe
3c = CN

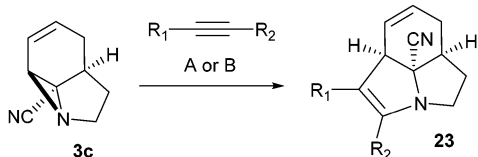
<p>10a 33%^[a] (22:1)^[b]</p>	<p>11a 88% (1:1)</p>	<p>12a 79% 12b 42% 12c <5%</p>
<p>13a 77% 13b 52%</p>	<p>14a 86%^[c] (1.8:1) 14b 87% (1.5:1)</p>	<p>15a 80% (4:1) 15b 78% (5.6:1) 15c 50% (7.4:1)</p>
<p>16a <5%</p>	<p>17a 28% (2.2:1)</p>	<p>18a 51%^[c] (3.1:1)</p>
<p>19a 64% (2:1)</p>	<p>20a 73% (4.2:1)</p>	<p>21a 47% (2.5:1)</p>
<p>22a 46% (7:1)</p>		

^a Required 64 h. ^b Diastereomeric ratio in parentheses. ^c 400 mol% of dipolarophile used.

reactive methylene diesters. However, use of nitrile aziridine **3c** gave lower yields of the corresponding products (*i.e.* **12c** & **15c**), with appreciable recovery of the starting material, due to the reduced reactivity of this substrate.

There have been a number of reports of the reaction of aldehydes with aziridines in a [3 + 2] manner, but with few exceptions¹¹ these all involved activated aziridines under Lewis acid catalysis.¹² We were intrigued to discover whether our strained aziridines would undergo [3 + 2] cycloadditions with aldehydes under the aforementioned optimised Pd-catalysed conditions. With the exception of benzaldehyde, a range of aldehydes reacted well with the aziridines to give a variety of



Table 4 [3 + 2] Cycloadditions of alkynes with cyanoaziridine **3c**


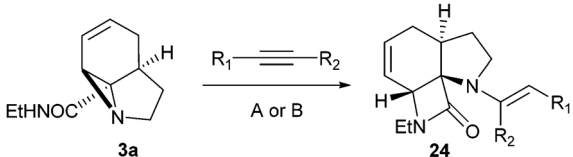
A. Pd₂(dba)₃ (3 mol%), P(OPh)₃ (0.25 eq), TBAI (0.1 eq), CH₂Cl₂ 36°C 16 h
B. MeCN, rt, 16 h

Entry	R ₁	R ₂	Conditions A yield (%)	Conditions B yield (%)
1	COMe	H	53	56
2	CHO	TMS ^a	60	62
3	CO ₂ Me	CO ₂ Me	92	94
4	CN	H	37	18

^a Desilylated product obtained on work-up (R₁ = CHO; R₂ = H).

functionalised oxazolidines with good diastereoselection (**13–18**). These results highlight the exceptional reactivity of these tricyclic aziridines compared to the more conventionally activated documented examples. It should be noted that none of these reactions gave oxazolidines in the absence of palladium (*vide infra*). Following on from this success we were keen to explore the use of imines in this reaction. Gratifyingly, it was found that a selection *N*-Ts imines reacted with **3a** to give the tricyclic fused aminals **19–22** with reasonable to good levels of diastereoselection. This appears to be a previously unreported mode of addition for aziridines, allowing novel access to highly functionalised aminals.¹³

Table 5 Formation of β-lactams by a novel Pd-catalysed addition/cyclisation sequence with alkynes



A. Pd₂(dba)₃ (3 mol%), P(OPh)₃ (0.25 eq), TBAI (0.1 eq), CH₂Cl₂, 36°C.
B. MeCN, rt.

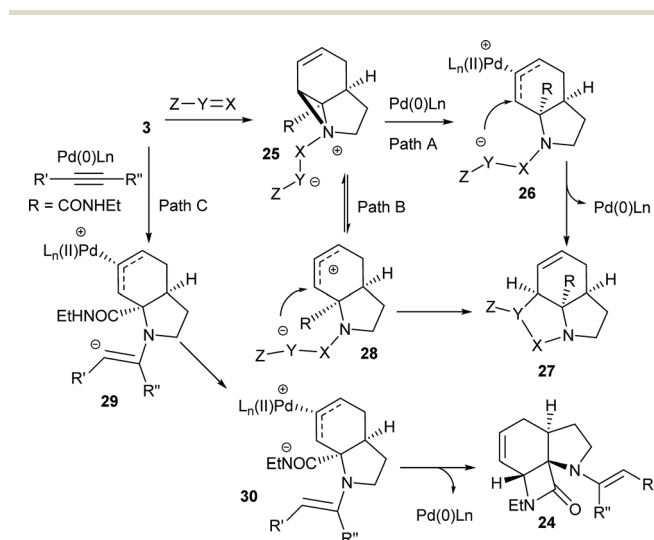
Entry	R ₁	R ₂	t (h)	Conditions A yield (%)	t (h)	Conditions B yield (%)
1	COMe	H	16	98	16	82
2	CO ₂ Me	H	16	83	24	54
3	CO ₂ Et	H	16	78	47	61
4	CONH ₂	H	16	0	16	0
5	CHO	TMS ^a	16	46 ^b	20	60 ^c
6	CO ₂ Me	CO ₂ Me	16	32	6.5	48

^a Desilylated product obtained on work-up (R₁ = CHO; R₂ = H). ^b 15% of an oxazolidinone by-product was also isolated – see ESI for details.

^c Includes 9% of the imide isomer – see ESI for details.

Finally, we decided to examine whether the limited scope of [3 + 2] cycloaddition reactions of alkenes (3 examples, Table 3) could be extended using alkyne based Michael acceptors. It was envisioned that the aziridines would react with alkynes under Pd-catalysis to generate tricyclic systems containing a partially unsaturated pyrrolidine ring. This proved successful in the reaction of **3c** with a range of alkynes and gave rapid access to the highly functionalised dihydropyrrole **23**. Interestingly, these cycloadditions proceeded equally well with or without catalyst (Table 4) suggesting, in this case, a purely thermal process. Moving to amide **3a** a quite different outcome was observed, and remarkably the tricyclic fused β-lactams **24** were obtained. This sequence produces a single diastereomeric product (Table 5). One notable feature here is that although Pd-catalysis generally gave the highest yields and shortest reaction times, the reactions were also found to proceed without catalyst at room temperature.

Although these reactions may seem diverse, mechanistically they are all likely to be related, with subtle differences depending on the substrates and conditions used. The unusual Tsuji–Trost type ring opening reactions described in Table 1 are likely to proceed *via* a conventional pathway where strain induces a higher reactivity compared to standard non-activated aziridines. It is possible, however, the aziridine is further activated by initial protonation from the carbon-nucleophiles used, which would not be possible for more conventionally activated aziridines (*e.g.* NTs). The reactions in Tables 2 and 3 can be described by the mechanisms in Scheme 4. Firstly the aziridine **3** reacts by nucleophilic attack on the isocyanate/ketone/aldehyde/imine to generate the zwitterion **25**. This is in contrast to initial attack of Pd to give a π-allyl Pd complex as previously described by Alper.^{8a} We propose this based on the absence of any Pd-induced reactions (*e.g.* ring opening/β-hydride elimination) when the electrophile is either not present or insufficiently reactive. Palladium then reacts with the zwitterion **25** to give π-allyl Pd complex **26** (Path A), which after cyclisation forms the



Scheme 4 Plausible mechanistic rationale for formation of all products.



various heterocycles **27** and regenerates the Pd(0) catalyst. Where the reaction proceeds without catalyst (*e.g.* Table 2, entry 18) it is possible that the zwitterion **25** is in equilibrium with the ring-opened form **28**, which then undergoes cyclisation to **27** (Path B). In the case of β -lactam formation it is likely that after the aziridine adds to the alkyne and undergoes ring opening with Pd(0), the resulting vinyl anion **29** deprotonates the secondary amide to give **30** which then cyclises to the 4-membered lactam (Path C). The higher pK_a of the vinyl anion **29** compared to other species (*i.e.* **26** & **28**) explains why β -lactam formation was not observed in other cases.

Conclusions

In summary, highly strained and compact fused tricyclic aziridines have been shown to undergo a range of ring-opening and cycloaddition reactions under mild conditions. This is in contrast with the usual behaviour of aziridines, which normally require an activating electron-withdrawing group to be placed on the nitrogen. Notable features include: (a) a very rare and extremely mild Tsuji–Trost type aziridine ring-opening using carbon-nucleophiles; (b) a Pd-catalysed [3 + 2] cycloaddition of aldehydes and imines, which has hitherto not been reported; (c) a novel, mild and effective method for the stereo-controlled formation of tricyclic fused β -lactams. Together these diverse reactions highlight the power of combining an initial photochemical step with a secondary catalytic process, in this case resulting in highly complex products in just two steps from simple pyrroles. We have also shown that the photochemical step to give these complex aziridines intermediates is readily scalable using the FEP flow reactors we have previously described,^{3a-c} allowing access to gram quantities of the substrates for these catalytic processes. As such these results may prove useful to chemists looking to exploit short routes to complex, sp^3 -rich compounds as potential scaffolds in drug discovery and alkaloid synthesis.

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Notes and references

- (a) J. B. Sweeney, *Chem. Soc. Rev.*, 2002, **31**, 247–258; (b) D. Tanner, *Angew. Chem., Int. Ed. Engl.*, 1994, **33**, 599–619; (c) For a review dealing specifically with vinyl aziridines see: H. Ohno, *Chem. Rev.*, 2014, **114**, 7784–7814.
- K. G. Maskill, J. P. Knowles, L. D. Elliott, R. W. Alder and K. I. Booker-Milburn, *Angew. Chem., Int. Ed.*, 2013, **52**, 1499–1502.
- (a) B. D. A. Hook, W. Dohle, P. R. Hirst, M. Pickworth, M. Berry and K. I. Booker-Milburn, *J. Org. Chem.*, 2005, **70**, 7558–7564; (b) L. D. Elliott, J. P. Knowles, P. J. Koovits, K. G. Maskill, M. J. Ralph, G. Lejeune, L. J. Edwards, R. I. Robinson, I. R. Clemens, B. Cox, D. D. Pascoe, G. Koch, M. Eberle, M. B. Berry and K. I. Booker-Milburn, *Chem.–Eur. J.*, 2014, **20**, 15226–15232; (c) J. P. Knowles, L. D. Elliott and K. I. Booker-Milburn, *Beilstein J. Org. Chem.*, 2012, **8**, 2025–2052; (d) S. Yuanha, K. Kuijpersa, V. Hessela and T. Noël, *Reaction Chemistry & Engineering*, 2016, DOI: 10.1039/c5re00021a; (e) A. C. Gutierrez and T. Jamison, *Org. Lett.*, 2011, **13**, 6414–6417; (f) A. Yavorsky, O. Shvydkiv, N. Hoffmann, K. Nolan and M. Oelgemoller, *Org. Lett.*, 2012, **14**, 4343–4345.
- (a) B. M. Trost, M. Osipov and G. Dong, *J. Am. Chem. Soc.*, 2010, **132**, 15800–15807; (b) S. Sebelius, V. J. Olsson and K. J. Szabó, *J. Am. Chem. Soc.*, 2005, **127**, 10478–10479.
- (a) A. A. Cantrill, A. N. Jarvis, H. M. I. Osborn, A. Ouadi and J. B. Sweeney, *Synlett*, 1996, 847–848; (b) T. Kawamura, N. Matsuo, D. Yamauchi, Y. Tanabe and H. Nemoto, *Tetrahedron*, 2013, **69**, 5331–5341.
- (a) B. M. Trost and T. R. Verhoeven, *J. Org. Chem.*, 1976, **41**, 3215–3216; (b) B. M. Trost and P. E. Strege, *J. Am. Chem. Soc.*, 1977, **99**, 1649–1651; (c) H. Matsushita and E. Negishi, *J. Chem. Soc., Chem. Commun.*, 1982, 160–161; (d) J. Fiaud and J. Legros, *J. Org. Chem.*, 1987, **52**, 1907–1911; (e) M. Luparia, M. T. Oliveira, D. Audisio, F. Frébault, R. Goddard and N. Maulide, *Angew. Chem., Int. Ed.*, 2011, **50**, 12631–12635.
- (a) K. Evstigneev and P. Maiorova, *Russ. J. Gen. Chem.*, 1997, **67**, 1475–1477; (b) K. Araki, K. Iwamoto, S. Shinkai and T. Matsuda, *Bull. Chem. Soc. Jpn.*, 1990, **63**, 3480–3485; (c) C. J. A. McCaw, A. O'Byrne and W. J. Spillane, *Eur. J. Org. Chem.*, 2008, **24**, 4200–4205; (d) D. Celadon, F. Maran, M. G. Severin and E. Vianello, *J. Am. Chem. Soc.*, 1991, **113**, 9320–9329.
- (a) D. C. D. Butler, G. A. Inman and H. Alper, *J. Org. Chem.*, 2000, **65**, 5887–5890; (b) B. M. Trost and D. R. Fandrick, *J. Am. Chem. Soc.*, 2003, **125**, 11836–11837.
- (a) K. Aoyagi, H. Nakamura and Y. Yamamoto, *J. Org. Chem.*, 2002, **67**, 5977–5980; (b) M. A. Lowe, M. Ostovar, S. Ferrini, C. C. Chen, P. G. Lawrence, F. Fontana, A. A. Calabrese and V. K. Aggarwal, *Angew. Chem., Int. Ed.*, 2011, **50**, 6370–6374; (c) K. Fugami, Y. Morizawa, K. Ishima and H. Nozaki, *Tetrahedron Lett.*, 1985, **26**, 857–860; (d) N. T. Patil and Y. Yamamoto, *Synlett*, 2007, 1994–2005.
- B. M. Trost and S. R. Angle, *J. Am. Chem. Soc.*, 1985, **107**, 6123–6124.
- (a) M. Hanaoka, A. Ashimori, H. Yamagishi and S. Yasuda, *Chem. Pharm. Bull.*, 1983, **31**, 2172–2175; (b) N. Murugesan, G. Blaskó, R. D. Minard and M. Shamma, *Tetrahedron Lett.*, 1981, **22**, 3131–3134; (c) M. Hanaoka, M. Kohzu and S. Yasuda, *Chem. Pharm. Bull.*, 1985, **33**, 4113–4115; (d) J. B. Dougherty, C. L. Lazzell and A. R. Collett, *J. Am. Chem. Soc.*, 1950, **72**, 2866–2867.
- (a) B. Kang, A. W. Miller, S. Goyal and S. T. Nguyen, *Chem. Commun.*, 2009, 3928–3930; (b) V. K. Yadav and V. Sriramurthy, *J. Am. Chem. Soc.*, 2005, **127**, 16366–16367; (c) S. Gandhi, A. Bisai, B. A. B. Prasad and V. K. Singh, *J. Org. Chem.*, 2007, **72**, 2133–2142; (d) M. K. Ghorai and K. Ghosh, *Tetrahedron Lett.*, 2007, **48**, 3191–3195; (e)



- R. Maeda, R. Ishibashi, R. Kamaishi, K. Hirotaki, H. Furuno and T. Hanamoto, *Org. Lett.*, 2011, **13**, 6240–6243; (f) H.-T. Yang, M.-L. Xing, Y.-F. Zhu, X.-Q. Sun, J. Cheng, C.-B. Miao and F.-B. Li, *J. Org. Chem.*, 2014, **79**, 1487–1492; (g) K. Okawa, T. Kinutani and K. Sakai, *Bull. Chem. Soc. Jpn.*, 1968, **41**, 1353–1355.
- 13 C–C bond thermolysis of aziridines generates azomethine ylides which undergo [3 + 2] cycloaddition reactions with imines: (a) A. Boruah, B. Baruah, D. Prajapati, J. S. Sandhu and A. C. Ghosh, *Tetrahedron Lett.*, 1996, **37**, 4203–4204; (b) M. T. Hancock and A. R. Pinhas, *Tetrahedron Lett.*, 2003, **44**, 7125–7128; (c) X. Wua and J. Zhang, *Synthesis*, 2012, **44**, 2147–2154.

